

Version: 4

Revision date: 2025-04-24

## Oxygen Bulk & Compressed

PG-SDS-08

This SDS conforms to the Globally Harmonised System (GHS), South African Regulations on Hazardous Chemical Agents, and SANS 10234, SANS 11014 & SANS 10228.

| SECTION 1: PRODUCT AND COMPANY IDENTIFICATION |   |  |
|---|---|--|
| Product name                                  | Oxygen - Instrument Grade, Instrument "O" Grade, Ultra High Purity  |  |
| Chemical Name                                 | Oxygen  |  |
| Other means of identification                 | Molecular oxygen, Oxygen compressed   |  |
| Recommended Intended Purpose                  | The product is used in various applications and may include - Technical processes, the Food Industry, and the Medical Industry. Always use as intended. |  |
| Company Information                           |   |  |
|   | Puregas (Pty) Ltd   |  |
|   | PO Box 123884, Alrode, 1451, Gauteng, South Africa  |  |
|   | Tel: (011) 903 9760   |  |
|   | Fax: (011) 903 9766   |  |
|   | Cellphone: 082 889 6946 (1st)   |  |
|   | 082 885 7475 (2 <sup>nd</sup> )   |  |
|   | Email: info@puregas.co.za   |  |
|   | Website: www.puregas.co.za  |  |
| Emergency Telephone                           |   |  |
|   | 0800 172 743  |  |
|   | Rapid Spill Response - 24 hours, 7 days a week  |  |

| CECTION 2 L                       | CECTION 2 HAZADDC IDENTIFICATION |   |   |   |
|-----------------------------------|----------------------------------|---|---|---|
| SECTION 2. HAZARDS IDENTIFICATION |                                  |   |   |   |
|                                   |                                  |   |   | published through ECHA  |
| Hazard Classificat                | ion                              | Category  | y Hazard Statement  |   |
| Oxidizing gases                   |                                  |   | H270  | May cause or intensify fire: oxidizer                           |
| Gases under press                 | sure (Comp.)                     | 2.2   | H280  | Contains gas under pressure; may explode if heated              |
|                                   |                                  | 2.2   | H281  | Contains refrigerated gas, may cause cryogenic burns or injury  |
| Hazard Pictogram                  | s                                |   | GHS   | 03 GHS04  |
| Signal Word                       |                                  | Danger  |   |   |
| Precautionary Sta                 | atements                         | l .   |   |   |
| General:                          | P101                             | If  | medical a   | dvice is needed, have product container or label at hand        |
|                                   |                                  |   | Keep out of reach of children                                       |   |
|                                   |                                  |   | ad carefu   | ully and follow all instructions                                |
| Prevention                        | P202                             |   |   | dle until all safety precautions have been read and understood. |
| P220<br>P244<br>P280              |                                  | Ke  | Keep/Store away from combustible materials, clothing                |   |
|                                   |                                  | Ke  | Keep reduction valves/valves and fittings free from oil and grease. |   |
|                                   |                                  | W   | Wear protective gloves/protective clothing/eye protection/face      |   |
|                                   |                                  | pr  | otection.   |   |
| Response                          | P370+P37                         | P370+P376 In  |   | re: Stop leak if safe to do so                                  |
| Storage                           | P271 + P403 Us                   |   | se and sto  | re only outdoors or in a well-ventilated place                  |
| Disposal                          |                                  | No  | None  |   |
| Main Hazard                       |                                  | Oxygen is non-flammable, but readily supports combustion. Never permit oil, grease or other readily combustible substance to come into contact with high concentrations of oxygen |   |   |



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| Flammability  | Not flammable, however, Oxidiser may intensify fires   |
|---------------|--|
| Health Hazard | Central Nervous system toxicity includes dizziness, convulsions, and loss of consciousness after only 2-3 hours of exposure to pure oxygen at 2 or more atmospheres e.g. sports and deep-sea diving. Essentially non-toxic   |
| Other Hazards | Adults can satisfactorily breathe pure oxygen for extended periods at 0.33 atm, or at 1 atm for several days at least 5 hours a day. However, irritation of mucous membranes may occur when 100% oxygen is inhaled continuously for several hours. Chest pains and cough can result from breathing O <sub>2</sub> at 1 atm for 8 to 24 hours or 2 atm for 2 to 3 hours or from an atmosphere of 60% oxygen for several days. A variety of central nervous system effects can occur from breathing oxygen at partial pressures greater than 2 atm, including dizziness, impaired coordination, visual and hearing disturbances, and seizures. Contact with liquid can cause severe frostbite/freeze burns. Prolonged breathing of very cold atmospheres can produce lung damage. Prolonged exposure to cold areas can result in hypothermia. Primarily entry: Inhalation. |

# SECTION 3. COMPOSITION/INFORMATION ON INGREDIENTS Substance CAS No BC No Name %

| CAS No                  | EC No     | Name                  | %          |
|-------------------------|-----------|-----------------------|------------|
| 7782-44-7               | 231-956-9 | Oxygen, Compressed    | 99.5 – 100 |
| UN Number - compressed: | 1072      | UN Number - liquified | 1073       |

See Section 8 for Exposure Guidelines and Section 15 for Regulatory Classifications.

## SECTION 4. FIRST AID MEASURES

In case of eye contact Not applicable. Non-irritating gas.

In case of skin contact Not applicable. Non-irritating gas.

In case of ingestion Ingestion is not an applicable route of exposure to gases

the victim to fresh air and obtain medical advice.

**Treatment (Advice to doctor)** Remove the victim to a warm (not hot) area. Remove contaminated clothing,

if possible. Wrap the person in blankets. Slowly restore body temperature.

Get medical help.

#### **SECTION 5. FIRE-FIGHTING MEASURES**

**Fire / Explosion hazard:**As oxygen is non-flammable, but strongly supports combustion, the correct type of extinguisher should be used depending on the combustible material

involved. If possible, shut off the source of escaping oxygen. All cylinders should be removed from the fire. Cylinders that cannot be removed should be

cooled with water from a safe distance

Special hazards arising from the

substance

Oxygen vigorously accelerates combustion. Materials that would not normally burn in air could combust vigorously in atmospheres having high

concentrations of oxygen.

Special protective equipment for

fire-fighters

Safety goggles, gloves, and safety shoes should be worn when handling

cylinders.



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#### **SECTION 6. ACCIDENTAL RELEASE MEASURES**

**Personal precautions** 

Workers handling liquid oxygen should wear safety glasses, clean approved insulated gloves, and other approved protective clothing as required to prevent skin contact. Gloves and protective clothing must be of a material that is resistant to ignition in contact with liquid oxygen; leather gloves and safety shoes have been recommended. Safety shoes and safety glasses are recommended when handling cylinders of compressed gas. Clothing that has been overexposed or contaminated with oxygen should be removed and considered unsafe (highly flammable) to wear for at least 30 minutes. If oxygen-enriched clothing catches fire, extinguish the flame under a safety shower; a fire blanket may not be effective in this situation. Use a continuous water spray to soak the clothing of a rescuer who must operate in an oxygen-enriched fire area.

**Environmental precautions** 

As gas is heavier than air, pockets of oxygen-enriched air could occur. These could lead to the fire spreading rapidly. If possible ventilate the affected area

Methods and material for containment and cleaning up

Notify safety personnel of leaks or spills. Evacuate all personnel from the danger area. Provide optimum exhaust ventilation. Shut off the source of the oxygen leak if you can do so without risk. Remove the source of heat, and ignition, and, if feasible, separate combustibles from the leak. Small leaks in an oxygen system in an enclosed, unventilated area can build up a hazardous oxygen level. To increase the rate of controlled evaporation of spilled liquid oxygen (when desired) spray the spill with large amounts of water (This may generate a fog and reduce visibility)

#### **SECTION 7. HANDLING AND STORAGE**

Precautions for safe handling and storage rooms and vessels

Do not allow cylinders to slide or come into contact with sharp edges. Cylinders of oxygen should not be stored near cylinders of acetylene or other combustible gases. Oxygen cylinders may be stacked horizontally provided that they are firmly secured at each end to prevent rolling. Prevent dirt, grit of any sort, oil, or any other lubricant from entering the cylinder valves, and store cylinders well clear of any corrosive influence e.g. battery acid. Compliance with all relevant legislation is essential. Use a "first in – first out" inventory system to prevent full cylinders from being stored for excessive periods of time Keep out of reach of children. NO SMOKING in areas of storage. Do not perform any welding, cutting, soldering, drilling, or other hot work on an empty vessel, container, or piping until all material has been cleared. No contact with incompatible materials such as oil and grease. Do not open the cylinder if damaged. Never use excessive force when opening. Make sure valves on gas cylinders are fully opened when gas is used. Open and shut the valve at least once a day, while the cylinder is in use to avoid "freezing". Make sure cylinders are labeled clearly. Do not lift cylinders by the cap or with a lifting magnet. Shut flow off at the cylinder valve and not just at the regulator after se. Regularly check cylinders for evidence of corrosion or leak. Keep empty cylinders under slightly positive pressure. Have suitable emergency equipment for fires, spills, and leaks readily available. Practice good housekeeping. Maintain handling equipment. Comply with applicable regulations.

## **SECTION 8. EXPOSURE CONTROLS / PERSONAL PROTECTION**

No exposure limits are available

Respiratory protection

No specific guidelines are available.



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| Hand protection          | Workers handling liquid oxygen should wear clean approved insulated gloves (Gloves must be of a material that is resistant to ignition on contact with liquid oxygen; leather gloves have been recommended)   |
|--------------------------|---|
| Eye protection           | Workers handling liquid oxygen should wear safety glasses.  |
| Skin and body protection | Workers should wear approved protective clothing as required to prevent skin contact. (Protective clothing must be of a material that is resistant to ignition on contact with liquid oxygen; leather safety shoes have been recommended) Clothing that has been overexposed or contaminated with oxygen should be removed and considered unsafe (highly flammable) to wear for at least 30 minutes. If oxygen-enriched clothing catches fire, extinguish the flame under a safety shower; a fire blanket may not be effective in this situation. Use a continuous water spray to soak the clothing of a rescuer who must operate in an oxygen-enriched fire area |
| Engineering Controls     | Where oxygen may be released, providing adequate ventilation to prevent excessive oxygen enrichment of the workplace atmosphere (holding at < 3% O2 by volume) is recommended for fire safety. Personnel who have been exposed to high concentrations of oxygen should stay in a well-ventilated or open area for 15 minutes before going into a confined space or near an ignition source.   |

### **SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES**

Physical state Colourless liquified gas Colour Colourless Odour odourless Taste **Tasteless** рΗ N/A Melting point - 219 °C **Boiling point** - 183 °C Auto-ignition temperature Not available Flammability (solid, gas) Not applicable Vapour pressure @ -199°C mmHg : ca 100 Density (Air = 1) 1.1Solubility: Water cm/100g @ 25°C: 3.16

(oil/water) : Log P(oct) = 0.65

### **SECTION 10. STABILITY AND REACTIVITY**

Solubility - coefficient

| Reactivity                         | No additional information available  |
|------------------------------------|--|
| Chemical Stability                 | Stable under normal conditions.  |
| Possibility of Hazardous reactions | Violently oxidizes organic material.   |
| Conditions to avoid                | The build-up of oxygen-enriched atmospheres, as, depending on temperature, oxygen reacts with all of the elements, except the inert gases, to form oxides These reactions can sometimes be violent, as with highly combustible materials such as oil and grease. Never use cylinders as rollers or supports or for any other purpose than the storing of oxygen. Never expose cylinders to excessive heat, as this may cause a sufficient build-up of pressure to rupture the cylinders. |
| Incompatible materials             | Since dry oxygen is non-corrosive most materials of construction are suitable.  Avoid all flammable materials.   |



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| Hazardous decomposition | Not known |
|-------------------------|-----------|
| products.               |           |

#### **SECTION 11. TOXICOLOGICAL INFORMATION**

#### Information on toxicological effects

Acute toxicity: Inhalation: Numero

Inhalation: Numerous animal studies demonstrate that high oxygen concentrations or pressures cause respiratory, central nervous system (CNS) and visual effects. Mortality in animals is generally related to lung damage and

pulmonarye dema

**Germ cell mutagenicity:** High oxygen concentrations at atmospheric pressure caused chromosomal

aberrations and mutations in Chinese hamster lung cells. A mutagenic effect of 70 to 95% oxygen was reported in Syrian hamster embryo fibroblasts. 95% oxygen/1% carbon dioxide induced chromosomal aberrations and sister

chromatid exchanges in Chinese hamster ovary cells.

Carcinogenicity: Not classified

Reproductive toxicity: Not classified

#### **SECTION 12. ECOLOGICAL INFORMATION**

No ecological damage was caused by this product.

#### **SECTION 13. DISPOSAL CONSIDERATIONS**

#### Waste treatment methods:

Remove waste containers or leaking cylinders to an open outdoor area away from combustibles and allow the oxygen to discharge at a moderate rate. Tag a leaking cylinder to indicate a defect, close the valve, and return the cylinder to its supplier

Packaging: The disposal of cylinders must only be handled by the gas supplier.

#### **SECTION 14. TRANSPORT INFORMATION**

#### **UN Pictogram**





#### Land and inland navigation transport ADR/RID

UN No. 1072 9compressed & UN No. 1073 (liquified), Shipping Name Oxygen, Class 2.2, Subsidiary Risk: Oxidizing agent, Hazchem Warning: 2.2 - Non-flammable gas, 5.1 - Oxidizer.

#### Marine transport IMDG

MDG 1072, Shipping Name Oxygen, Class 2.2, Subsidiary Risk: Oxidizing agent, Label: Non-flammable, non-toxic gas.

#### Air transport ICAO/IATA-DGR

ICAO/IATA Code 1072, Class 2.2, Packing Group: - Packaging instructions - Cargo: allowed - Passenger: allowed

#### Special precautions for user

The protective measures listed in Sections 6, 7, and 8 of the Safety Data Sheet have to be considered.

#### **SECTION 15. REGULATORY INFORMATION**

Safety, health, and environmental regulations/legislation specific to the substance or mixture:

Occupational Health and Safety Act, Hazardous Chemical Agents Regulations

SANS 11014:2010 Edition 1



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SANS 10228:2012 Edition 6 SANS 10234:2019 Edition 2 SUPPLEMENT TO SANS 10234 Edition 1 National Road Traffic Act Dangerous Goods Regulations

#### **SECTION 16. OTHER INFORMATION**

#### SELECTED BIBLIOGRAPHY

- 1. Data sheets as supplied by various Suppliers and Manufacturers
- 2. Emergency Response Handbook Annex A of SABS 0232-3
- 3. GHS Purple booklet
- 4. ECHA
- 5. Occupational Health and Safety Act, Hazardous Chemical Agents Regulations
- 6. SANS 11014:2010 Edition 1
- 7. SANS 10228:2012 Edition 6
- 8. SANS 10234:2019 Edition 2
- 9. SUPPLEMENT TO SANS 10234 Edition 1
- 10. National Road Traffic Act
- 11. Dangerous Goods Regulations

#### R/S- phrases:

- R8 Contact with combustible material may cause fire
- R37 Irritating to respiratory system
- R48 Danger of serious damage to health by prolonged exposure
- S2 Keep out of reach of children
- S9 Keep container in a well-ventilated place
- S15 Keep away from heat
- S21 When using do not smoke
- S27 Take off immediately all contaminated clothing

#### Disclaimer:

This information is based on our current knowledge and is intended to describe the product for the purposes of health, safety, and environmental requirements only. It should not therefore be construed as guaranteeing any specific property of the product.