

**SAFETY DATA SHEET**Revised edition no : 2  
Date : 12/05/2014**Hydrogen Gas (All grades)**

PG 017

**SECTION 1. IDENTIFICATION OF THE PRODUCT AND COMPANY UNDERTAKING**

<b>Trade Name</b>	Hydrogen Gas (All Grades)
<b>Chemical Family</b>	N/A
<b>Chemical Name</b>	Hydrogen Gas
<b>Synonyms</b>	Hydrogen, Hydrogen, compressed, Molecular hydrogen
<b>Chemical Abstract no</b>	1333-74-0
<b>NIOSH No</b>	N/A
<b>UN no</b>	1049
<b>Company:</b>	<b>Puregas (Pty) Ltd</b> PO Box 123884, Alrode, 1451, South Africa <b>Tel :</b> (011) 903 9760 <b>Fax:</b> (011) 903 9766 <b>Cellphone:</b> 082 889 6946 (1 <sup>st</sup> ) 082 885 7475 (2 <sup>nd</sup> ) Info@puregas.co.za <b>Emergency Tel:</b> 0800 172 743 (Rapid Spill Response)

**SECTION 2. HAZARDS IDENTIFICATION**

<b>Main Hazard</b>	The main health hazard associated with releases of this gas is asphyxiation, by displacement of oxygen. The cryogenic liquid will rapidly boil to the gas at standard temperatures and pressures. The liquefied gas can cause frostbite to any contaminated tissue.
<b>Flammability</b>	Flammable Gas (burns at all ambient temperatures)
<b>Chemical hazard</b>	An extreme explosion hazard exists in areas in which the gas has been released but the material has not yet ignited. Hydrogen burns with an almost invisible blue flame
<b>Biological hazard</b>	Currently, biological exposure indices (BEIS) are not applicable for this product
<b>Reproduction hazard</b>	This product is not expected to cause adverse reproductive effects in humans.
<b>Eye effects</b>	Contact with rapidly expanding gases can be irritating to exposed eyes
<b>Health effects – skin</b>	Contact with cryogenic liquid or rapidly expanding gases (which are released under high pressure) may cause frostbite. Symptoms of frostbite include change in skin colour to white or grayish-yellow. The pain after contact with liquid can quickly subside
<b>Health effects – ingestion</b>	Not applicable to gases
<b>Health effects – inhalation</b>	High concentrations of this gas can cause an oxygen-deficient environment. Individuals breathing such an atmosphere may experience symptoms which include headaches, ringing in ears, dizziness, drowsiness, unconsciousness, nausea, vomiting and depression of all the senses. Under some circumstances of over-exposure, death may occur.  The following effects associate with various levels of oxygen are as follows: 12-16% oxygen – breathing and pulse rate increased, muscular co-ordination

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	slightly disturbed 10-14% - emotional upset, abnormal fatigue, disturbed respiration 6-10% - nausea and vomiting, collapse or loss of consciousness Below 6% - convulsive movements, possible respiratory collapse, and death
<b>Carcinogenicity</b>	No known effect
<b>Mutagenicity</b>	No known effect
<b>Neurotoxicity</b>	No known effect

**SECTION 3. COMPOSITION/INFORMATION ON INGREDIENTS**

<b>Hazardous components</b>	May contain small amounts of helium and hydrocarbon gas such as methane
<b>EEC Classification</b>	N/A
<b>R Phrases</b>	N/A

**SECTION 4. FIRST AID MEASURES**

<b>Eye contact</b>	Not applicable
<b>Skin contact</b>	In case of frostbite, place the frostbitten part in warm water. Do not use hot water. If warm water is not available, or is impractical to use, wrap the affected parts gently in blankets. Alternatively, if the fingers or hands are frostbitten, place the affected area in the armpit. Encourage victim to gently exercise the affected part while being warmed. Seek medical attention.
<b>Ingestion</b>	Not applicable
<b>Inhalation</b>	Rescuers should not attempt to retrieve victims of exposure to this product without adequate personal protective equipment. At a minimum, self-contained breathing apparatus and fire retardant personal protective equipment should be worn. Adequate fire protection must be provided during rescue situations. Remove victim(s) to fresh air, as quickly as possible. Trained personnel should administer supplemental oxygen and/or cardio-pulmonary resuscitation, if necessary. Only trained personnel should administer supplemental oxygen.

**SECTION 5. FIRE-FIGHTING MEASURES**

<b>Extinguishing Media</b>	Dry chemical, carbon dioxide, water spray, foam, fog
<b>Special Hazards</b>	Container may explode in fire.
<b>Protective Clothing</b>	Wear suitable protective clothing
<b>Firefighting instructions</b>	Do not extinguish a leaking gas flame unless leak can be securely plugged. Stop flow of gas and move containers from fire area if without risk. Use water to keep fire-exposed containers cool. Container may explode in fire.



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### SECTION 6. ACCIDENTAL RELEASE MEASURES

<b>Personal Precautions</b>	Restrict access to area until completion of clean-up. Ensure clean-up is conducted by trained personnel only. Wear adequate personal protective equipment. Extinguish or remove all ignition sources. Ventilate area
<b>Environmental Precautions</b>	No data available
<b>Small spills / Large spills</b>	Stop leak if it can be done without risk. Use water spray to reduce vapour. Isolate area until gas has dispersed

### SECTION 7. HANDLING AND STORAGE

<b>Suitable material</b>	N/A
<b>Handling/storage precautions</b>	<p>Do not use near welding operations, flames or hot surfaces. Move cylinders by hand-truck or cart designed for that purpose. Do not lift cylinders by their caps. Do not handle them with oily hands. Secure cylinder in place in an upright position at all times. Do not drop cylinders or permit them to bang against each other. Leave valve cap on cylinder until cylinder is secured and ready for use. Close all valves when not in actual use. Use smallest possible amounts in designated areas with adequate ventilation. Have emergency equipment (for fires, spills, leaks, etc.) readily available</p> <p>Comply with all applicable regulations for the storage and handling of compressed gases and flammable materials.</p> <p>Store at or above ground level, in a cool, dry, well-ventilated area, out of direct sunlight and away from heat and ignition sources. Cylinder temperature should never exceed 51°C. Label empty cylinders. Store full cylinders separately from empty ones. Empty containers may be hazardous due to residual material. Limit quantity of material in storage. Restrict access to storage area. Post appropriate warning signs. Keep storage area separate from populated work areas. Consider leak detection and alarm equipment for storage area</p>

### SECTION 8. EXPOSURE CONTROLS / PERSONAL PROTECTION

<b>Occupational exposure limits.</b>	No data available
<b>Engineering control measures</b>	Engineering control methods to reduce hazardous exposures are preferred. Methods include mechanical ventilation (dilution and local exhaust), process or personnel enclosure, control of process conditions and process modification (e.g. substitution of a less hazardous material). Administrative controls and personal protective equipment may also be required. Provide adequate local exhaust and dilution (general) ventilation to control airborne hydrogen below 4000 ppm (10% of the lower explosive limit). Exhaust directly to the outside. Supply sufficient replacement air to make up for air removed by exhaust systems.
<b>Personal protection – respiratory</b>	Maintain oxygen levels above 19.5% in the workplace. Use supplied air respiratory protection if oxygen levels are below 19.5% or during emergency response to a release of this product.
<b>Personal protection – hand.</b>	Wear mechanically-resistant gloves when handling cylinders of this product. Use low-temperature protective gloves (i.e. kevlar) when working with containers of liquid hydrogen.



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<b>Personal protection – eye</b>	Splash goggles or safety glasses, for protection from rapidly expanding gases and splashes of liquid hydrogen
<b>Personal protection – skin</b>	Use body protection appropriate for task. Transfer of large quantities under pressure may require protective equipment appropriate to protect employees from splashes of liquefied product, as well as fire retardant items
<b>Other protection</b>	None required

## SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES

<b>Appearance:</b> Colourless gas	<b>Explosive properties:</b> None
<b>Odor:</b> Odourless	<b>Oxidizing properties:</b> None
<b>pH:</b> N/A	<b>Vapour pressure :</b> N/A
<b>Boiling point:</b> -252.8°C	<b>Density:</b> 0.0695 (air = 1)
<b>Melting point:</b> -259.2°C	<b>Solubility - water:</b> Slightly soluble (1.8% v/v at 20°C)
<b>Flash point:</b> Flammable gas (burns at all ambient temperatures)	<b>Solubility - solvent:</b> Slightly soluble in ethanol, ether
<b>Flammability:</b> Lower 4.0% Upper 75%	<b>Solubility - coefficient:</b> N/A
<b>Auto flammability:</b> 571.2°C	

## SECTION 10. STABILITY AND REACTIVITY

<b>Conditions to Avoid</b>	Contact with incompatible materials and exposure to heat, sparks and other sources of ignition. Cylinders exposed to high temperatures or direct flame can rupture or burst
<b>Incompatible materials</b>	Strong oxidizers (i.e. chlorine, bromine, pentafluoride, oxygen, oxygen difluoride, and nitrogen trifluoride). Oxygen/hydrogen mixtures can explode on contact with a catalyst such as platinum
<b>Hazardous decomposition products</b>	Hydrogen. When ignited in the presence of oxygen, water will be produced

## SECTION 11. TOXICOLOGICAL INFORMATION

<b>Acute toxicity</b>	There is no specific toxicological data for hydrogen. Hydrogen is a simple asphyxiant, which acts to displace oxygen in the environment
<b>Chronic toxicity</b>	Irritancy of product : Contact with rapidly expanding gases can be irritating to exposed skin and eyes
<b>Carcinogenicity</b>	No data available
<b>Mutagenicity</b>	Not suspected to be a cancer-causing agent
<b>Reproductive hazards</b>	This product is not expected to cause mutagenic effects in humans  This product is not expected to cause adverse reproductive effects in humans

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**SECTION 12. ECOLOGICAL INFORMATION**

Hydrogen occurs naturally in the atmosphere. The gas will be dissipated rapidly in well-ventilated areas

Any adverse effect on animals would be related to oxygen deficient environments. No adverse effect is anticipated to occur to plant-life, except for frost produced in the presence of rapidly expanding gases

No evidence is currently available on this product's effects on aquatic life

<b>Aquatic toxicity – fish</b>	As above
<b>Aquatic toxicity – daphnia</b>	As above
<b>Aquatic toxicity – algae</b>	As above
<b>Biodegradability</b>	As above
<b>Bio-accumulation</b>	As above
<b>Mobility</b>	As above
<b>German wgk</b>	N/A

**SECTION 13. DISPOSAL CONSIDERATIONS**

<b>Disposal methods</b>	Waste disposal must be in accordance with appropriate local regulations. Return cylinders with any residual product to supplier. Do not dispose of locally
<b>Disposal packaging</b>	N/A

**SECTION 14. TRANSPORT INFORMATION**

<b>UN no:</b> 1049/1966	<b>IMDG - EMS No:</b> N/A
<b>ADR/RID Substance identity no:</b> 1049/1966	<b>IMDG – MFAG table on :</b> N/A
<b>ADR/RID class:</b> 2.1 Flammable gas	<b>IATA - class :</b> 2.1 Flammable gas
<b>ADR/RID item no:</b> N/A	<b>IATA – subsidiary risk(s):</b> N/A
<b>ADR/RID hazard identity no:</b> N/A	<b>UK – description:</b> No data available
<b>IMDG - shipping name:</b> (1049) Hydrogen, compressed or Hydrogen (1966) Hydrogen, refrigerated liquid	<b>UK emergency action class:</b> No data available
<b>IMDG - class:</b> 2.1 Flammable gas	<b>UK classification :</b> No data available
<b>IMDG – packaging group:</b> N/A	<b>Tremcard No:</b> N/A
<b>IMDG - marine pollutant:</b> Not classified as a marine pollutant	



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### SECTION 15. REGULATORY INFORMATION

**EEC hazard classification :** F+ - Extremely flammable

**Risk phrases :** Extremely flammable (R12)

**Safety phrases:** Keep out of reach of children. Keep container in a well-ventilated place, Keep away from sources of ignition – No smoking. Take precautionary measures against static discharges (S2- 9-16-33)

**National Legislation :** *OSH Act, National Road Traffic Act (when promulgated)*

### SECTION 16. OTHER INFORMATION

#### SELECTED BIBLIOGRAPHY

1. *Data sheets as supplied by various Suppliers and Manufacturers*
2. *Emergency Response Handbook - Annex A of SABS 0232-3*
3. *Handling Chemicals Safety, 2<sup>nd</sup>. Ed. Dutch Association of Safety Experts, Dutch Chemical Industry Association, Dutch Safety Institute, 1980*
4. *NIOSH Pocket Guide to Chemical Hazards, NIOSH, June 1990*
5. *Micromedex, Inc. TOMES CPS™ System Vol. 39*
6. *Patty's Industrial Hygiene Toxicology, 4<sup>th</sup> ed. Vol. II Part A, George D Clayton, Florence E Clayton*
7. *Supplement to NIOSH Manual of Analytical Methods, 3<sup>rd</sup> ed., NIOSH Publication No 84-100, 1985*
8. *Toxic & Hazardous Industrial Safety Manual - Industrial Chemicals Safety Manual for Handling and disposal with toxicity and hazard data*
9. *WinSpirs 2.1 as supplied by the Canadian Centre for Occupational Health and Safety.*

***All reasonable care has been exercised in processing your request for information on the chemical listed in this Material Safety Data Sheet. No warranty is made, either express or implied and Perisseuma Projects does not hold itself liable for any injury, illness, loss or misinterpretation arising from the use of this data.***